



PATENT

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Dated: September 27, 2007

by:

Rodney D. DeKruif

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of: Emrick et al.

Serial No: 10/643,015

Filed: August 18, 2003

For: PYRIDINE AND
RELATED LIGAND
COMPOUNDS,
FUNCTIONALIZED
NANOPARTICULATE
COMPOSITES AND
METHODS OF
PREPARATION

Attorney Docket No. 7163

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450

RULE 131 DECLARATION OF HABIB SKAFF

1. I, Habib Skaff, am a co-inventor with regard to the invention (the "Invention") disclosed and claimed in the above-entitled application (the "Application"). I make this declaration in support of the Application and, in particular, to antedate a reference cited against the Application.

2. The Invention claimed in the Application was completed before the effective date of application serial number 10/219,440 (*i.e.*, the Dubertret

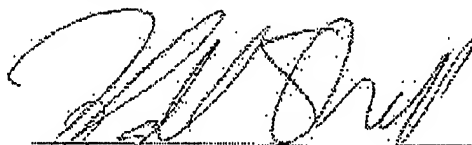
reference). More specifically, the Invention was conceived and with due diligence reduced to practice, in this country--the United States of America, prior to the effective date of the Dubertret reference.

3. This Declaration, and prior invention, is supported by copies of pertinent pages from my laboratory research notebook, entries to which I contemporaneously signed and dated and were witnessed by co-inventor, Todd S. Emrick. Date redacted copies of the aforementioned notebook pages are provided collectively as Exhibit A and incorporated herein by reference. These documents establish that the Invention was made at least as early as June 1, 2002, which is a date earlier than the effective date of the Dubertret reference. Without limitation, facts demonstrating prior invention of a composite of independent claim 1 include the experimental data I entered on page 37 of Exhibit A. Facts demonstrating prior invention of a system of independent claim 14 include the experimental data I entered on page 37 of a Exhibit A. Facts demonstrating prior invention of a method of independent claim 20 include the experimental data I entered on page 38 of Exhibit A.

I hereby declare that: All statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; that those statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under section 1001 of Title 18 of the United States Code; and that willful false

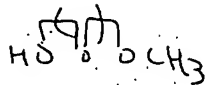
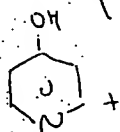
statements may jeopardize the validity of the Application or any patent issuing
thereon.

Date 9/29/07

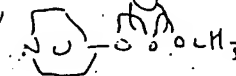

Habib Skaff

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DIAD

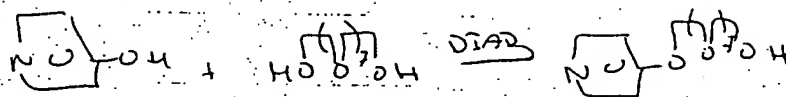
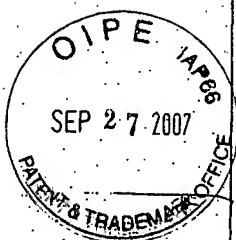
Reagents

- 95 ① 2g, 0.022 mol
 251 ② m-Py 750 14.25g, 0.019 mol
 262 ③ Ph₃P 6.25g, 0.024 mol
 222 ④ DIAD 4.84g, 0.024 mol (4.72 mL)
 ⑤ THF (dry) 300 mL 250 mL

Procedure

- ① Ph₃P + THF loaded into 2-neck flask & stirred under N₂ @ r.t.
- ② DIAD added via syringe & stirred for 1/2 hr.
- ③ phenol & alcohol added & stirred
- ④ reacted overnight
- ⑤ removed off THF
- ⑥ added DIAD & ether → washed w/ ether
- ⑦ extracted product out w/ CH₂Cl₂ out of AA phase → MgSO₄, Rotavap
- pmr show some [?] → triggered redissolving in d₂o with base (1.5M NaOH) & precipitating into CH₂Cl₂ (cold) were
- can column elute w/ CH₂Cl₂ : MeOH (7:3:0), (7:2:1)

Exhibit A



Reagents

450 ① $\text{N} \begin{array}{|c|} \hline \text{O} \\ \hline \end{array} \text{OH}$ & 2g, 0.011 mol

400 ② $\text{HO} \begin{array}{|c|} \hline \text{O} \\ \hline \end{array} \text{OH}$ 22g, 0.055 mol
p = 1.03

202 ③ DIAO 2.63g, 2.55 ml 0.013 mol

262 ④ Ph_3P 3.41g, 0.03

⑤ $\text{THF}(\text{dry})$ 300 mL

Procedure

① Ph_3P + THF loaded into 3-neck 500 mL round bottom
stirrer @ rt under N_2

② DIAO added via syringe & stirred for 1 hr

③ phenol & Et_2O added & stirred

→ reacted over night

- rotated off all THF

→ note

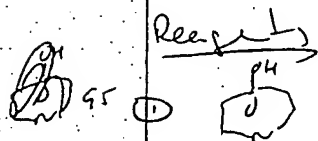
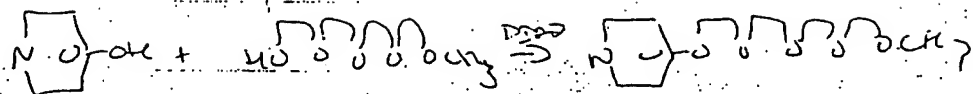
- extracted w/ H_2O → then aqueous wash

w/ CH_2Cl_2 → too difficult to purify by column

→ rotated off CH_2Cl_2 → dissolved in H_2O ,

washed w/ ether, then Toluene → doesn't work well if

try ~~acidify~~ acidifying aqueous to make pyridine salt
which will not be soluble in



5g, 0.055g mol

128. (c) m-Ty

5.632g, 0.044 mol

— 262 (3) Ph, P

13.1g 0.05 mol

202 (472740)

10.1 g, 0.05 mol, 9.85 mL

⑤ HF (aq)

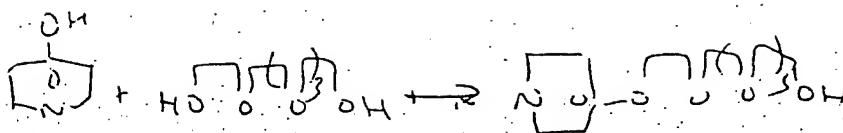
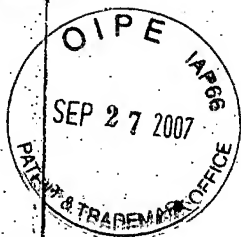
~~250 mL~~ 400 mL

Procedure

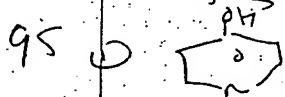
~~Ph₃P~~ Ph₃P : THF loaded into 2-neck flask
& stirred under N₂ a.s.e.

② DDA added in 5g & stirred for 1/2 hr.

③ phenol + alcohol added + stirred overnight



Reagents



4g, 0.042 mol

300 Hex

31.58g, 0.105 mol

262 Ph_3P

13.1g, 0.045

202

10.1g, 0.05 mol, 9.85 mL

THF

500 mL

Procedure

1) phenol, Ph_3P , DIAD, & THF loaded in 2-neck & stirred @ rt under N_2 for 1/2 hr.

2) diol added \Rightarrow stirred overnight

3) removed all THF

4) hexane/ CHCl_3 @ CHCl_3 : H_2O (20:20) 3) CHCl_3 : H_2O (75:20:5)

5) run column eluting w/ 1) CHCl_3 : H_2O (7:2:1)

6) stirred distilling off unreacted diol @ 224

@ 600 mtorr \Rightarrow didn't work well

7) run column in CHCl_3 : H_2O (75:20:5), (75:20:5), (80:20:10)

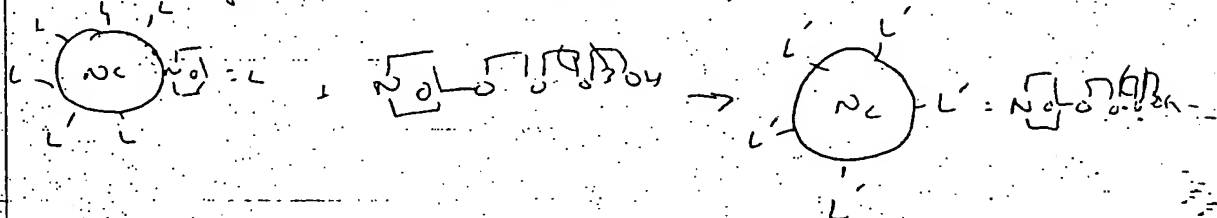
Paul J. Allen

John E. Blum

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John E. Blum

Exchange to hex-methoxydimethyl ether



Reagent

- ① pyridine M ~ 40mg
- ② $\text{hex-methoxydimethyl ether}$ 600mg
- ③ THF (dry) 3mL
- ④ DIW 6mL

Procedure

A) ① 20mg M dispersed in solution at 300mg
new ligand in THF \rightarrow immediately went
it into solution

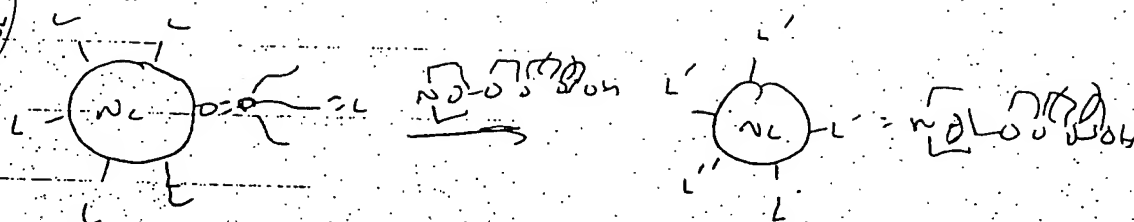
② dried under N_2 flow and added 3mL
DIW \rightarrow most went into solution \rightarrow centrifuged

B) ③ 20mg M dispersed in solution at 300mg
new ligand in 3mL DIW \rightarrow it went into
solution \rightarrow centrifuged *Juniper de Smeere*

John Hall

John Hall

K.H. Be



② TOPSO covered NC ~ 15mg
④ $\sqrt{2} - 0.5$ 320mg
③ THF (2mg) 3mL

- ① Ni made as ^{usual} usually & washed w/
MeOH 3 times
- ② dried over N_2 flow
- ③ redissolved in new ligand in THF and
allowed to stand over head of N_2 overnight
- ④ distilled at $1/2$ THF \rightarrow precipitated w/
hexane \rightarrow all Ni precipitated
- ⑤ washed w/ hexanes \rightarrow centrifuged \rightarrow
redissolved in THF (1)

Bill Hall

K. H. B. L.

Lab

Truett & Green